Eplerenone tetrahydrofuran solvate

The eplerenone molecule of the title compound, C_{24}H_{30}O_{6} \cdot C_{4}H_{8}O, contains three six-membered rings, two five-membered rings and one three-membered ring. In the crystal structure, the eplerenone molecules are linked via C—H⋯O hydrogen bonds, forming a 303 Å³ cavity, which is occupied by two disordered solvent THF molecules.

Comment

Eplerenone is known as an aldosterone receptor antagonist (Grob et al., 1985). Recently, we prepared single crystals of eplerenone from a tetrahydrofuran (THF) solution and determined the crystal structure of the title compound, (I).

The eplerenone molecule contains three six-membered rings, two five-membered rings and one three-membered ring (Fig. 1). Both five-membered rings display a similar envelope configuration. In the O3-containing ring, atom C18 lies at the flap position and is displaced by 0.386 (8) Å from the mean plane formed by the other four atoms; in the C15-containing ring, C13 lies at the flap position and is displaced by 0.683 (6) Å from the mean plane formed by the other four atoms. The C2=C3 bond distance (Table 1) shows typical C=C double-bond character.

Atom O5 of the C23-containing ester group forms an intramolecular C—H⋯O hydrogen bond with the C14-methine group (Fig. 1 and Table 2); the C9—C8—C23—O5 torsion angle is 72.7 (5)°. The C7—C8—C23—O6 torsion angle is 16.2 (5)°, leading to a shorter O6⋯C7 contact of 2.706 (6) Å. These structural features are identical with those found in the crystal structure of eplerenone dichloromethane solvate (Grob et al., 1997).

In the crystal structure, adjacent eplerenone molecules are linked to each another via C—H⋯O hydrogen bonds involving the carbonyl groups (Table 2), forming a 303 Å³ cavity (Spek, 2003) which is occupied by two disordered THF solvent molecules (Fig. 2).
Experimental

A microcrystalline powder sample of eplerenone was prepared in the manner reported by Grob et al. (1997). Single crystals of (I) were obtained from a THF solution of eplerenone.

Crystal data

\[ C_{24}H_{30}O_6 \cdot C_4H_8O \]

\[ M_r = 486.58 \]

Orthorhombic, \( P2_12_12_1 \)

\[ a = 8.2974 (7) \ \text{Å} \]

\[ b = 13.1068 (11) \ \text{Å} \]

\[ c = 23.3855 (15) \ \text{Å} \]

\[ V = 2560.7 (3) \ \text{Å}^3 \]

\[ Z = 4 \]

Mo Kα radiation

\[ \mu = 0.09 \ \text{mm}^{-1} \]

\[ T = 295 (2) \ \text{K} \]

\[ 0.32 \times 0.19 \times 0.13 \ \text{mm} \]

Data collection

Rigaku RAXIS-RAPID IP diffractometer

Absorption correction: none

24665 measured reflections

3304 independent reflections

2006 reflections with \( I > 2\sigma(I) \)

\[ R_{int} = 0.053 \]

Refinement

\[ R[F^2 > 2\sigma(F^2)] = 0.070 \]

\[ wR(F^2) = 0.229 \]

\[ S = 1.07 \]

3304 reflections

308 parameters

10 restraints

H-atom parameters constrained

\[ \Delta p_{	ext{min}} = 0.31 \ \text{e Å}^{-3} \]

\[ \Delta p_{	ext{max}} = -0.35 \ \text{e Å}^{-3} \]

Table 1

<table>
<thead>
<tr>
<th>Selected bond lengths (Å).</th>
</tr>
</thead>
<tbody>
<tr>
<td>O1—C1 1.241 (7)</td>
</tr>
<tr>
<td>O3—C20 1.346 (5)</td>
</tr>
<tr>
<td>O3—C17 1.487 (5)</td>
</tr>
<tr>
<td>O4—C20 1.209 (6)</td>
</tr>
</tbody>
</table>

Table 2

<table>
<thead>
<tr>
<th>Hydrogen-bond geometry (Å, °).</th>
</tr>
</thead>
<tbody>
<tr>
<td>D—H—A</td>
</tr>
<tr>
<td>D—H</td>
</tr>
<tr>
<td>H—A</td>
</tr>
<tr>
<td>D—A</td>
</tr>
<tr>
<td>D—H—A</td>
</tr>
<tr>
<td>C11—H11···O4i 0.98 2.60 3.410 (6) 140</td>
</tr>
<tr>
<td>C12—H12B···O1ii 0.97 2.55 3.499 (7) 168</td>
</tr>
<tr>
<td>C14—H14···O5 0.98 2.48 3.096 (6) 120</td>
</tr>
<tr>
<td>C19—H19B···O1vi 0.97 2.58 3.240 (7) 125</td>
</tr>
<tr>
<td>C21—H21A···O3vi 0.96 2.55 3.344 (7) 141</td>
</tr>
</tbody>
</table>

Symmetry codes:

(i) \(-x+1, y-z, -z+\frac{1}{2}\)

(ii) \(-x, y+\frac{1}{2}, -z+\frac{1}{2}\)

(iii) \(-x+1, y-z, -z+\frac{1}{2}\)

An initial difference Fourier map indicated that the solvent THF molecule is disordered, and a two-site disorder model was adopted in the refinement. Isotopic refinement for the two disordered components gave occupancies of 0.521 (9) and 0.479 (9), respectively. Geometrical restraints were used, and the displacement parameters of some pairs of atoms were kept equal to each other. The H atoms of the disordered THF molecules were not located. Methyl H atoms were placed in calculated positions, C—H = 0.96 Å, and torsion angles were refined to fit the electron density; \( U_{	ext{rot}}(H) = 1.5U_{	ext{eq}}(C) \). Other H atoms were placed in calculated positions, C—H = 0.93–0.98 Å, and refined in riding mode with \( U_{	ext{rot}}(H) = 1.2U_{	ext{eq}}(C) \). In the absence of significant anomalous scattering effects, Friedel pairs were merged; the absolute configuration of (I) was not determined.

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2002); program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

Figure 1

The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). The minor disorder component of the THF solvent molecule has been omitted for clarity. The dashed line indicates the intramolecular hydrogen bond.

Figure 2

The packing of (I). H atoms and the minor disorder component of THF have been omitted for clarity.
The work was supported by the Natural Science Foundation of China (grant No. 20443003).

References


